

Abundance variables first derivatives - 12Aug2017

The composition derivatives are perhaps more nuanced than one might initially imagine because one of them is special.

Baryon number is an invariant. Define the abundance of species Y_i by

$$Y_i = \frac{n_i}{n_B} = \frac{N_i}{N_B} \quad (1)$$

where N_i is the number of particles of isotope i , N_B is the number of baryons, n_i is the number density [cm^{-3}] of isotope i and n_B is baryon number density [cm^{-3}]. Define the average of any quantity $\bar{\beta}$ by the number density n_i weighted average

$$\bar{\beta} = \frac{\sum \beta_i n_i}{\sum n_i} \quad (2)$$

with equation (1) becomes

$$\bar{\beta} = \frac{\sum \beta_i Y_i}{\sum Y_i} . \quad (3)$$

The partial derivative with respect to abundance Y_i is

$$\frac{\partial \bar{\beta}}{\partial Y_i} = \frac{\beta_i}{\sum Y_i} - \frac{\sum \beta_i Y_i}{(\sum Y_i)^2} = \frac{\beta_i}{\sum Y_i} - \frac{\bar{\beta}}{\sum Y_i} = \frac{\beta_i - \bar{\beta}}{\sum Y_i} . \quad (4)$$

In this case,

$$\begin{aligned} \frac{\partial \bar{A}}{\partial Y_i} &= \frac{A_i - \bar{A}}{\sum Y_i} \\ \frac{\partial \bar{Z}}{\partial Y_i} &= \frac{Z_i - \bar{Z}}{\sum Y_i} \end{aligned} \quad (5)$$

However, reconsider the special case when the arbitrary quantity β_i is the number of nucleons A_i . As shown earlier, baryon conservation requires $\sum X_i = \sum A_i Y_i = n_B/n_B = 1$. Then $\bar{A} = 1/\sum Y_i$ and another valid expression for $\partial \bar{A}/\partial Y_i$ is

$$\begin{aligned} \frac{\partial \bar{A}}{\partial Y_i} &= -\bar{A}^2 \\ \frac{\partial \bar{Z}}{\partial Y_i} &= \bar{A} (Z_i - \bar{Z}) . \end{aligned} \quad (6)$$

Despite looking very different, equations (5) and (6) are equivalent, provided one **consistently** uses either set of expressions. Mixing the two formalisms, essentially when to set $\sum X_i = 1$, causes confusion and incorrect results. Let's show the equivalence for the two forms of $d\bar{A}/dY_i$.

The ideal gas

$$P = \left(\sum n_i \right) kT = \left(\sum Y_i \right) n_B kT , \quad (7)$$

has the trivial first derivative with respect to abundance

$$\frac{\partial P}{\partial Y_i} = n_B kT \quad (8)$$

Consider the formalism of equation (5) where $\sum X_i = 1$ was not explicitly invoked. The ideal gas law, using equation (5) becomes

$$P = n_B kT \left(\sum Y_i \right) = n_B kT \left(\frac{\sum A_i Y_i}{\bar{A}} \right) . \quad (9)$$

The derivative with respect to abundance is

$$\begin{aligned} \frac{\partial P}{\partial Y_i} &= n_B kT \left[\frac{A_i}{\bar{A}} - \frac{\sum A_i Y_i}{\bar{A}^2} \frac{\partial \bar{A}}{\partial Y_i} \right] \\ &= n_B kT \left[\frac{A_i}{\bar{A}} - \frac{\sum A_i Y_i}{\bar{A}^2} \left(\frac{A_i - \bar{A}}{\sum Y_i} \right) \right] \\ &= n_B kT \left[\frac{A_i}{\bar{A}} - \frac{A_i \sum A_i Y_i}{\bar{A}^2 \sum Y_i} + \frac{1}{\bar{A}} \frac{\sum A_i Y_i}{\sum Y_i} \right] \\ &= n_B kT \left[\frac{A_i}{\bar{A}} - \frac{A_i}{\bar{A}} + 1 \right] \\ &= n_B kT \end{aligned} \quad (10)$$

which is identical to equation (8).

Consider the formalism of equation (6) where $\sum X_i = 1$ is used. The ideal gas law becomes

$$P = n_B kT \left(\sum Y_i \right) = \frac{n_B kT}{\bar{A}} . \quad (11)$$

The derivative with respect to abundance is

$$\frac{\partial P}{\partial Y_i} = \frac{\partial P}{\partial \bar{A}} \frac{\partial \bar{A}}{\partial Y_i} = \left[-\frac{n_B kT}{\bar{A}^2} \right] \left[-\frac{1}{\bar{A}} \right] = n_B kT \quad (12)$$

which is identical to equation (8).

This shows either formalism may be used, provided it is used consistently. Since many expressions that use \bar{A} are written in the form of equation (11) rather than equation (9), the first derivative form of equation (6) is preferred.