

## Abundance variables first derivative examples - 12Aug2017

We've previously shown the mean atomic number

$$\bar{A} = \frac{\sum n_i A_i}{\sum n_i} = \frac{\sum Y_i A_i}{\sum Y_i} = \frac{1}{\sum Y_i} \quad (1)$$

and mean charge

$$\bar{Z} = \frac{\sum n_i Z_i}{\sum n_i} = \frac{\sum Y_i Z_i}{\sum Y_i} = \bar{A} \sum Y_i Z_i \quad (2)$$

have derivatives

$$\frac{d\bar{A}}{dY_i} = -\bar{A}^2 \quad (3)$$

and

$$\frac{\partial \bar{Z}}{\partial Y_i} = \bar{A} (Z_i - \bar{Z}) \quad (4)$$

respectively. For an arbitrary quantity  $\beta$  that is written in terms of  $\bar{A}$  and  $\bar{Z}$ , the derivative with respect to  $Y_i$  is

$$\frac{d\beta}{dY_i} = \frac{\partial \beta}{\partial \bar{Z}} \frac{\partial \bar{Z}}{\partial Y_i} + \frac{\partial \beta}{\partial \bar{A}} \frac{\partial \bar{A}}{\partial Y_i} = \frac{\partial \beta}{\partial \bar{Z}} \bar{A} (Z_i - \bar{Z}) - \frac{\partial \beta}{\partial \bar{A}} \bar{A}^2. \quad (5)$$

One assumes all partials of  $\beta$  with respect to  $\bar{A}$  and  $\bar{Z}$  are available from the physics is at hand (e.g., from an eos).

One simple but relevant example is the specific energy of an ideal gas

$$e = \frac{3P}{2\rho} = \frac{3N_A kT}{2\bar{A}} \quad \text{erg g}^{-1} \quad (6)$$

has the partial derivative of with respect to  $\bar{Z}$

$$\frac{\partial e}{\partial \bar{Z}} = 0 \quad (7)$$

and the partial derivative of with respect to  $\bar{A}$

$$\frac{\partial e}{\partial \bar{A}} = -\frac{3N_A kT}{2\bar{A}^2}. \quad (8)$$

Hence by equation (5) the derivative with respect to  $Y_i$  is

$$\frac{de}{dY_i} = \frac{\partial e}{\partial \bar{Z}} \frac{\partial \bar{Z}}{\partial Y_i} + \frac{\partial e}{\partial \bar{A}} \frac{\partial \bar{A}}{\partial Y_i} = \left[ -\frac{3N_A kT}{2\bar{A}^2} \right] \left[ -\bar{A}^2 \right] = \frac{3}{2} N_A kT \quad \text{erg g}^{-1} \quad (9)$$

as expected when recasting equation (6) as

$$e = \frac{3P}{2\rho_B} = \frac{3}{2} N_A kT \sum Y_i \quad \text{erg g}^{-1}. \quad (10)$$

Note  $\partial e / \partial Y_i$  is usually known as the chemical potential in the first law of thermodynamics when composition changes are taken into account.

Another example is the mean molecular weight

$$\mu = \left[ \frac{1}{\mu_{ion}} + \frac{1}{\mu_{ele}} \right]^{-1} = \left[ \frac{1}{\bar{A}} + Y_e \right]^{-1} = \left[ \frac{1}{\bar{A}} + \frac{\bar{Z}}{\bar{A}} \right]^{-1} = \frac{\bar{A}}{\bar{Z} + 1} = \frac{n_B}{\sum n_i + n_e}. \quad (11)$$

The partial derivative of  $\mu$  with respect to  $\bar{Z}$  is

$$\frac{\partial \mu}{\partial \bar{Z}} = \frac{\partial}{\partial \bar{Z}} \left( \frac{\bar{A}}{\bar{Z} + 1} \right) = -\frac{\bar{A}}{(\bar{Z} + 1)^2} = -\frac{\mu}{\bar{Z} + 1} = -\frac{\mu^2}{\bar{A}}. \quad (12)$$

The partial derivative of  $\mu$  with respect to  $\bar{A}$  is

$$\frac{\partial \mu}{\partial \bar{A}} = \frac{\partial}{\partial \bar{A}} \left( \frac{\bar{A}}{\bar{Z} + 1} \right) = \frac{1}{(\bar{Z} + 1)} = \frac{\mu}{\bar{A}} \quad (13)$$

Applying equation (5), the derivative of  $\mu$  with respect to  $Y_i$  is

$$\begin{aligned} \frac{d\mu}{dY_i} &= \frac{\partial \mu}{\partial \bar{Z}} \frac{\partial \bar{Z}}{\partial Y_i} + \frac{\partial \mu}{\partial \bar{A}} \frac{\partial \bar{A}}{\partial Y_i} = \left[ -\frac{\mu^2}{\bar{A}} \right] [\bar{A}(Z_i - \bar{Z})] + \left[ \frac{\mu}{\bar{A}} \right] [-\bar{A}^2] \\ &= \mu^2(\bar{Z} - Z_i) - \mu \bar{A} \end{aligned} \quad (14)$$

Sometimes the composition is defined in terms of the hydrogen baryon (mass) fraction  $\mathcal{X}$ , helium baryon (mass)  $\mathcal{Y}$ , and metal baryon (mass)  $\mathcal{Z}$ . Baryon number conservation is then expressed as

$$\mathcal{X} + \mathcal{Y} + \mathcal{Z} = 1. \quad (15)$$

These abundance variables are related to the usual  $Y_i = X_i A_i$  abundance variables by

$$\mathcal{X} = Y_H \quad \mathcal{Y} = 4Y_{He} \quad \mathcal{Z} = A_z Y_Z, \quad (16)$$

which mostly explains the use of calligraphic font to avoid the same symbols for different quantities. Applying equation (1)

$$\bar{A} = \frac{1}{\mathcal{X} + \mathcal{Y}/4 + \mathcal{Z}/A_z} = \frac{4A_z}{A_z \mathcal{X} + A_z \mathcal{Y} + 4\mathcal{Z}} = \frac{4A_z}{A_z + 3A_z \mathcal{X} + 4\mathcal{Z} - A_z \mathcal{Z}}, \quad (17)$$

where the last expression eliminated  $\mathcal{Y}$  using equation (15). Applying equation (2)

$$\begin{aligned} \bar{Z} &= \frac{\mathcal{X} + 1/2\mathcal{Y} + Z_z/A_z \mathcal{Z}}{\mathcal{X} + \mathcal{Y}/4 + \mathcal{Z}/A_z} \\ &= (\mathcal{X} + 1/2\mathcal{Y} + Z_z/A_z \mathcal{Z}) \cdot \bar{A} \\ &= \frac{2A_z(1 + \mathcal{X} - \mathcal{Z}) + 4Z_z \mathcal{Z}}{A_z + 3A_z \mathcal{X} + 4\mathcal{Z} - A_z \mathcal{Z}}, \end{aligned} \quad (18)$$

where again  $\mathcal{Y}$  was eliminated using equation (15).

Then the first derivatives of  $\bar{A}$  are

$$\begin{aligned}\frac{\partial \bar{A}}{\partial \mathcal{X}} &= -\frac{12A_z^2}{(A_z + 3A_z\mathcal{X} + 4\mathcal{Z} - A_z\mathcal{Z})^2} \\ \frac{\partial \bar{A}}{\partial \mathcal{Z}} &= \frac{4A_z^2 - 16A_z}{(A_z + 3A_z\mathcal{X} + 4\mathcal{Z} - A_z\mathcal{Z})^2}\end{aligned}\quad (19)$$

The first derivatives of  $\bar{Z}$  are

$$\begin{aligned}\frac{\partial \bar{Z}}{\partial \mathcal{X}} &= \frac{4A_z[A_z(\mathcal{Z} - 1) + \mathcal{Z}(2 - 3Z_z)]}{(A_z + 3A_z\mathcal{X} + 4\mathcal{Z} - A_z\mathcal{Z})^2} \\ \frac{\partial \bar{Z}}{\partial \mathcal{Z}} &= -\frac{4A_z[2 + \mathcal{X}(2(A_z - 3Z_z) - Z_z)]}{(A_z + 3A_z\mathcal{X} + 4\mathcal{Z} - A_z\mathcal{Z})^2}\end{aligned}\quad (20)$$

For an arbitrary quantity  $\beta$  that is written in terms of  $\bar{A}$  and  $\bar{Z}$ , the derivatives with respect to  $\mathcal{X}$  and  $\mathcal{Z}$  are

$$\begin{aligned}\frac{d\beta}{d\mathcal{X}} &= \frac{\partial \beta}{\partial \bar{Z}} \frac{\partial \bar{Z}}{\partial \mathcal{X}} + \frac{\partial \beta}{\partial \bar{A}} \frac{\partial \bar{A}}{\partial \mathcal{X}} \\ \frac{d\beta}{d\mathcal{Z}} &= \frac{\partial \beta}{\partial \bar{Z}} \frac{\partial \bar{Z}}{\partial \mathcal{Z}} + \frac{\partial \beta}{\partial \bar{A}} \frac{\partial \bar{A}}{\partial \mathcal{Z}}\end{aligned}\quad (21)$$

Note that equations (17) and (18) can be solved for the either unknown or inconvenient variables  $A_z$  and  $Z_z$ :

$$\begin{aligned}A_z &= \frac{4\bar{A}\mathcal{Z}}{4 - \bar{A} - 3\bar{A}\mathcal{X} - \bar{A}\mathcal{Z}} \\ Z_z &= \frac{4\bar{Z} - 2\bar{A}(1 + \mathcal{X} - \mathcal{Z})}{4 - \bar{A} - 3\bar{A}\mathcal{X} - \bar{A}\mathcal{Z}}.\end{aligned}\quad (22)$$

Eliminating  $A_z$  and  $Z_z$  from equation (19) gives

$$\begin{aligned}\frac{\partial \bar{A}}{\partial \mathcal{X}} &= -\frac{3\bar{A}^2}{4} \\ \frac{\partial \bar{A}}{\partial \mathcal{Z}} &= \frac{\bar{A}(\bar{A} + 3\bar{A}\mathcal{X} - 4)}{4\mathcal{Z}}.\end{aligned}\quad (23)$$

Eliminating  $A_z$  and  $Z_z$  from equation (20) gives

$$\begin{aligned}\frac{\partial \bar{Z}}{\partial \mathcal{X}} &= \frac{\bar{A}(2 - 3\bar{Z})}{4} \\ \frac{\partial \bar{Z}}{\partial \mathcal{Z}} &= \frac{\bar{A}[\bar{Z} + (3\bar{Z} - 2)\mathcal{X} - 2]}{4\mathcal{Z}}.\end{aligned}\quad (24)$$

Equation (21) still holds.