

## Reaction Networks - 12Aug2017

Baryon number is an invariant. Define the abundance of species  $Y_i$  by

$$Y_i = \frac{n_i}{n_B} = \frac{N_i}{N_B} \quad (1)$$

where  $N_i$  is the number of particles of isotope  $i$ ,  $N_B$  is the number of baryons,  $n_i$  is the number density [ $\text{cm}^{-3}$ ] of isotope  $i$  and  $n_B$  is baryon number density [ $\text{cm}^{-3}$ ]. The number of baryons in isotope  $i$  divided by the total number of baryons is the baryon fraction  $X_i$ ,

$$X_i = Y_i A_i = \frac{n_i A_i}{n_B} \quad (2)$$

where  $A_i$  is the atomic mass number, the number of baryons in an isotope. Usually the baryon fraction is called the “mass fraction”. Note

$$\sum X_i = \frac{n_B}{n_B} = 1 \quad (3)$$

is invariant under nuclear reactions. Define the baryon density, in atomic mass units, as

$$\rho_B = n_B m_u = \frac{n_B}{N_A} \text{ g cm}^{-3} \quad (4)$$

where  $m_u$  is the atomic mass unit [g] and  $N_A$  is the Avogadro number [ $\text{g}^{-1}$ ] in a system of units where the atomic mass unit is *defined* as 1/12 mass of an unbound atom of  $^{12}\text{C}$  is at rest and in its ground state.

The continuity equation for the number density of species  $i$  in an Eulerian framework is

$$\frac{\partial n_i}{\partial t} + \frac{\partial(n_i v_x)}{\partial x} = \sum_{j,k} r_{jk} n_j n_k \quad \text{cm}^{-3} \text{ s}^{-1} \quad (5)$$

where the reaction rate between two species  $j$  and  $k$  is

$$r_{jk} = \langle \sigma v \rangle_{jk} \quad \text{cm}^3 \text{ s}^{-1} \quad (6)$$

and  $\langle \sigma v \rangle_{jk}$  is the cross-section  $\sigma$  [in  $\text{cm}^2$ ] times the relative speed  $v$  [in  $\text{cm s}^{-1}$ ] between the two isotopes, and the angled brackets indicates an average over a statistical distribution, usually a Maxwell-Boltzmann.  $r_{jk}$  is a function of temperature only. The reaction rate implies a lifetime for isotope  $j$  of  $\tau_j = 1/(n_j r_{jk})$  s.

Nuclear reactions, and expansion or contraction of the plasma can produce changes in the number densities  $n_i$ . To separate the nuclear changes in composition from hydrodynamic effects,

substituting equation (1) gives

$$\begin{aligned} \frac{\partial(Y_i n_B)}{\partial t} + \frac{\partial(Y_i n_B v_x)}{\partial x} &= \sum_{j,k} r_{j,k} Y_j Y_k n_B^2 \\ n_B \frac{\partial Y_i}{\partial t} + Y_i \frac{\partial n_B}{\partial t} + n_B \frac{\partial(Y_i v_x)}{\partial x} + Y_i \frac{\partial(n_B v_x)}{\partial x} &= \sum_{j,k} r_{j,k} Y_j Y_k n_B^2 \\ n_B \left( \frac{\partial Y_i}{\partial t} + \frac{\partial(Y_i v_x)}{\partial x} \right) + Y_i \left[ \frac{\partial \rho}{\partial t} + \frac{\partial(\rho v_x)}{\partial x} \right] &= \sum_{j,k} r_{j,k} Y_j Y_k n_B^2 \end{aligned} \quad (7)$$

The term in square brackets is zero by the mass continuity equation. Thus,

$$\frac{\partial Y_i}{\partial t} + \frac{\partial(Y_i v_x)}{\partial x} = \sum_{j,k} r_{j,k} n_B Y_j Y_k \quad \text{s}^{-1} \quad (8)$$

or in a Lagrangian frame

$$\frac{dY_i}{dt} = \sum_{j,k} r_{j,k} n_B Y_j Y_k = \sum_{j,k} r_{j,k} N_A \rho Y_j Y_k \quad \text{s}^{-1} \quad (9)$$

In an operator split Eulerian hydrocode, the advection term is done separately, leading to the same ordinary differential equations to solve as in the Lagrangian form.

Common reaction rate compilations list  $\lambda = N_A \langle \sigma v \rangle_{jk}$  [in  $\text{cm}^3 \text{g}^{-1} \text{s}^{-1}$ ], so

$$\frac{dY_i}{dt} = \sum_{j,k} \lambda_{j,k} \rho Y_j Y_k \quad \text{s}^{-1} \quad (10)$$

Let  $R_{j,k} = \lambda_{j,k} \rho$  be the ‘‘reaction rate’’ that subsumes all the temperature and density dependences. Then,

$$\frac{dY_i}{dt} = \sum_{j,k} R_{j,k} Y_j Y_k \quad \text{s}^{-1} \quad (11)$$

are the equations that constitute a nuclear reaction network.